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Approved for public release; distribution unlimited.

13. SUPPLEMENTARY NOTES

AFOSR Ionic Liquids Workshop

Tampa, FL, 7-8 March 2004

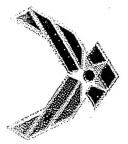
14. ABSTRACT

20040503 194

15. SUBJECT TERMS

16. SECURITY CLASSIFICATION OF:			17. LIMITATION OF ABSTRACT	18. NUMBER OF PAGES	19a. NAME OF RESPONSIBLE PERSON Linda Talon
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IL QC QSPR - Preliminary Results

Jeffrey D. Mills Jeffrey.Mills@edwards.af.mil

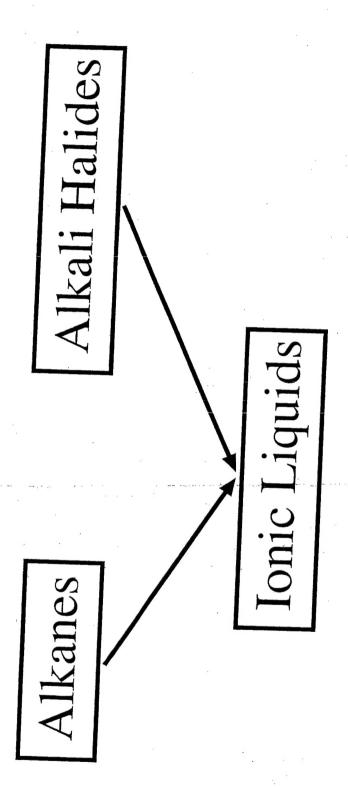
AFRL/PRSP

Space and Missile Propulsion Division Air Force Research Laboratory Propellants Branch

QSPR



Quantitative Structure-Property Relationships Property(Descriptor)



Property(Descriptor)



Melting Point
Density (liquid)
(Decomp. Temp.)
(Impact Sensitivity)
(Density (Solid))
(Others ??)

QC Descriptors:

- Single Particle:

Surface Electrostatic Potential Surface Size and Shape

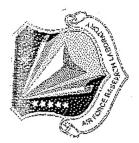
- Pair:

Separation Binding Energy





Constraints/Goals



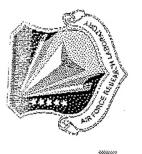
1. Predictive (not just summarize or interpolate)

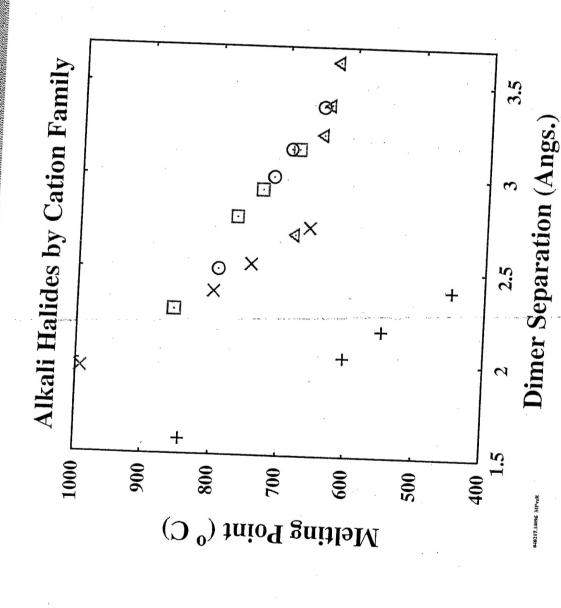
"Universal" Descriptors (Ionic and Nonionic)

3. Allow Ion Interchangablility

4. Physically Meaningful and Chemically Ignorant a. No "Kitchen-Sink" Fits b. Charge Symmetry

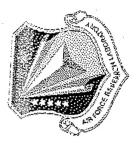
Cation Families Summarized



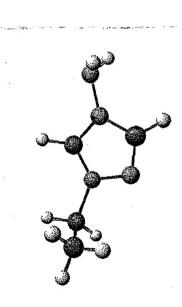


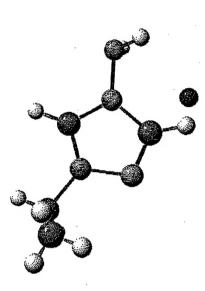




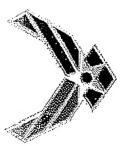


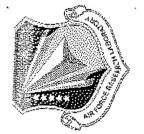
B3LYP 6-311++G(d,p): Crystal (with Br⁻): 1-Ethyl-4-amino-1,2,4-triazolium Cation











Color Map:

V at 0.001 au Electron Density:

kcal/mol

350.0

291.7

233,3

Na+:

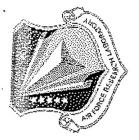


 $\mathbb{R}\mathrm{b}^+$:

175.0







kcal/mol

350.0

300.0

250.0

200.0

150.0

100.0

50.0

0.0

-50.0

-100.0

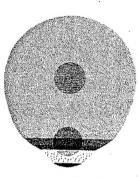
-150.0

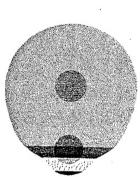
-250.0

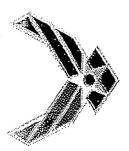
-300.0

-350.0

-200.0

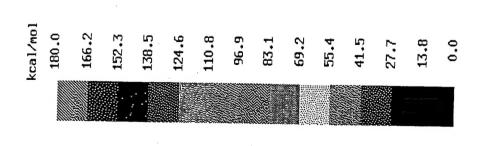


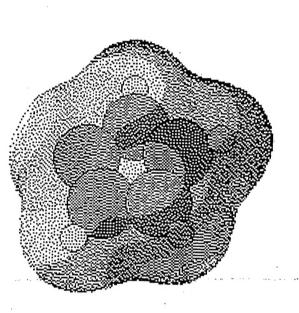


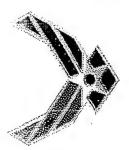




1,2,4-Triazolium Cation









1-R_n-4-amino-1,2,4-triazolium Cation, n=even:

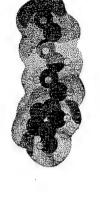
n=0:

kcal/mol

166.2

152.3

n=6:



n=8:

n=2:

124.6

110.8

6.96

83.1

69.2

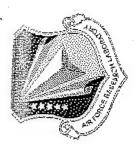
55.4



n=4:



Descriptors



Electrostatic Potential:

$$\Pi = rac{1}{\mathrm{A}} \sum_{\mathrm{i}} \left| \mathrm{V_i} - ar{\mathrm{V}} \right| \mathrm{A_i}$$

Size

$$\mathrm{A} = \sum_{\mathrm{i}} \mathrm{A}_{\mathrm{i}}$$

Shape:

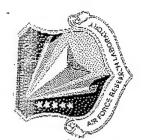
$$(I_0, I_1, I_2)$$
 - PA Moments of Extensia (ordered)

Asphericity =
$$\frac{I_2 - I_0}{\bar{r}}$$

Blateness =
$$\frac{I_1 - I_0}{I_2 - I_0}$$

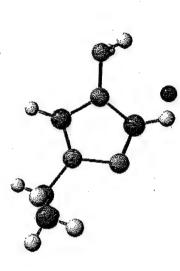


Separation and Interaction Energy



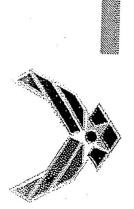
1-Ethyl-4-amino-1,2,4-triazolium Bromide B3LYP 6-31+G(d):

Crystal:

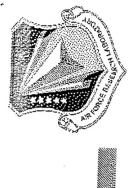


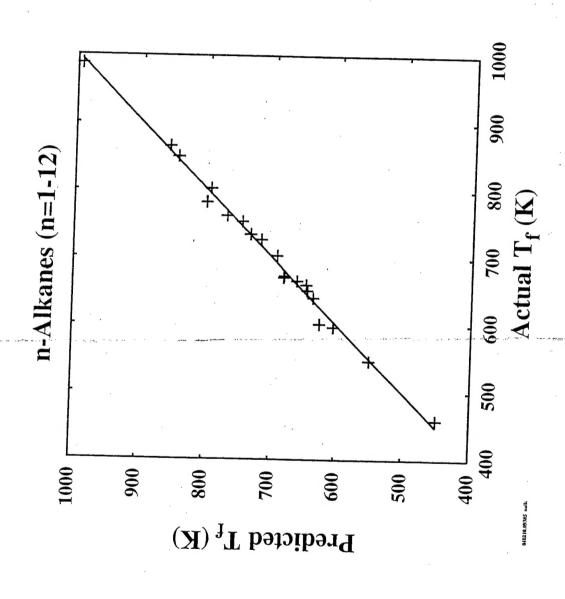
Binding Energy





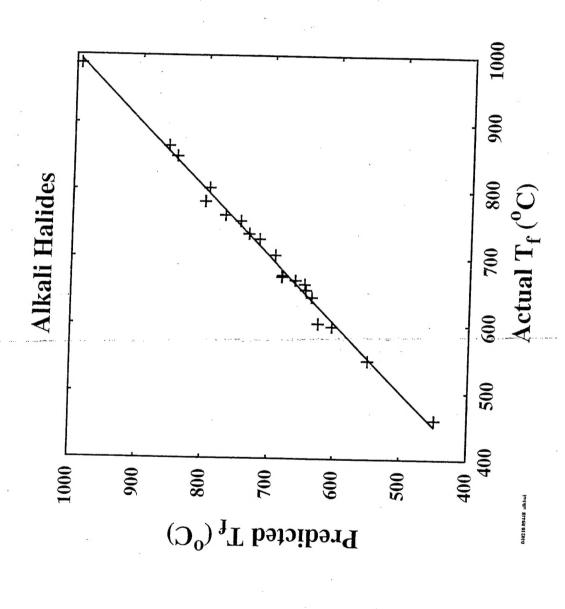
QSPR Correlation





QSPR Correlation



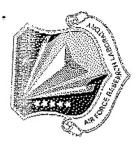


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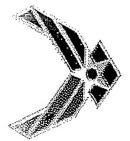


Concluding (Open) Question



Can QC QSPR aid the IL synthesist or will it merely follow?





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G. Drake, L. Hall

